EFFECT OF THE ELECTROMAGNETIC COMPOUND FIELD ON THE GRAIN GROWTH AND WEAR RESISTANCE OF A Cu-Pb MONOTECTIC ALLOY

VPLIV SESTAVLJENIH PULZIRAJOČIH IN ELEKTROMAGNETNIH POLJ NA RAST KRISTALNIH ZRN IN ODPORNOST Cu-Pb ENO-EVTEKTIČNE ZLITINE PROTI OBRABI

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Prejem rokopisa – received: 2019-05-29; sprejem za objavo – accepted for publication: 2019-09-30

doi:10.17222/mit.2019.115

The Cu-37.4w/%Pb (Cu-Pb) monotectic alloys were prepared under the normal solidification conditions and intense-pulse/ electromagnetic-compound fields. The Cu-Pb monotectic alloy prepared under the normal solidification conditions was mainly composed of α -Cu and severely segregated Pb. With the compound fields applied, the segregation in the Cu-Pb alloy fabrication process could be eliminated. The alloy was composed of α -Cu and β -Pb, and its crystals were lath-shaped with much finer grains compared to the one fabricated under the normal conditions. In addition, the crystal grains grew along the tangential direction of the pulse current radiationally, from the guiding center to the surrounding area, caused by the Lorentz-force-induced gyroscopic motion. From the friction and wear experiments, the Cu-Pb alloy prepared by compound fields possesses a higher hardness, a lower friction coefficient and better wear resistance.

Keywords: intense pulse and electromagnetic compound fields, Cu-37.4w/%Pb, Pb monotectic alloy, grain growth, wear resistance

Avtorji članka so pripravili eno-evtektično zlitino Cu-37,4 mas. % Pb (Cu-Pb) v normalnih pogojih strjevanja in v pogojih intenzivnih sestavljenih pulzirajočih elektromagnetnih polj. Eno-evtektična Cu-Pb zlitina, pripravljena v normalnih pogojih strjevanja, je bila v glavnem sestavljena iz α -Cu in močno segregiranega Pb. Pri uporabi sestavljenih polj so odpravili segregacije Pb med izdelavo Cu-Pb zlitina, je bila v glavnem sestavljeni z α -Cu in močno segregiranega Pb. Pri uporabi sestavljenih polj so odpravili segregacije Pb med izdelavo Cu-Pb zlitina, je bila sestavljena iz homogenih trdnih raztopin α -Cu in β -Pb. Kristalna zrna so imela obliko letvic z mnogo bolj finimi kristalnimi zrni v primerjavi s tisto mikrostrukturo zlitine, ki je bila izdelana v normalnih pogojih strjevanja. Dodatno so ugotovili, da so kristalna zrna rasla vzdolž tangencialne smeri širjenja tokovnih pulzov s središčem vodenega preseka okolice, povzročenih z Lorenzovo silo induciranega žiroskopskega gibanja. Preizkusi trenja in obrabe so pokazali, da ima Cu-Pb zlitina, pripravljena v sestavljenih poljih, večjo trdoto in manjši koeficient trenja ter boljšo odpornost proti obrabi.

Ključne besede: intenzivna sestavljena pulzirajoča in elektromagnetna polja, Cu-37,4 mas. % Pb, eno-evtektična zlitina, rast kristalnih zrn, odpornost proti obrabi

1 INTRODUCTION

As one of the important metal materials in engineering, monotectic alloys have many unique properties and are widely used.¹⁻⁷ Cu-Pb is a typical monotectic alloy, which has a high bearing capacity, a low friction coefficient and excellent wear/corrosion resistance. It has been widely used in thermal conduction and wearresistant materials. However, the phase separation easily occurs during alloy preparation with normal solidification due to the considerably large difference between the density of Cu and Pb. Thus strong segregation of the secondary phase and obvious stratification phenomenon exist in the Cu-Pb monotectic alloy, which results in substantial deterioration.

Based on an understanding of biphasic separation mechanism, many new methods have been attempted to prepare monotectic alloys with excellent properties. Abramov prepared Al-Pb anti-friction alloys by casting in a microgravity field. It was found that the aluminum matrix produced a uniform distribution and possesses high anti-friction properties.8 E. G. Wang and L. Zhang,9 made research on the solidification structure of a Cu-80wt%Pb alloy, which was investigated under four different experimental conditions. The result showed that a magnetic field has a remarkable effect on restraining the gravity segregation of the Cu-Pb alloy. Y. Liu analyzed the liquid phase separating mechanism of immiscible alloys and summarized the synthetic techniques of homogeneous immiscible alloys.¹⁰ W. X. Hao¹¹ prepared Cu-37.4w/%Pb alloys in the highly undercooling conditions. It was found that the grain of the alloy was gradually refined with the increase of the cooling rate and the alloy showed a good anti-friction property. M.

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Induction heating coil

Figure 1: Schematic diagram of our own designed equipment

Zhu¹² prepared Al-Pb-Cu alloy by mechanical alloying (MA), the result revealed the strength of the alloy was increased owing to solid-solution strengthening and precipitation strengthening. In addition, unidirectional solidification, stirring casting and rapid solidification are also widely used in the preparation of monotectic alloys.¹⁰⁻¹⁵

In this paper the Cu-37.4w/%Pb (Cu-Pb) monotectic alloy was prepared by the intense pulse and electromagnetic compound fields, where the undercooling and magnetostriction influence on the Cu-Pb monotectic alloy grain growth was investigated. Furthermore, the influence of the compound fields on the composition separation of the Cu-Pb monotectic alloy was analyzed and described. The refinement mechanism of the crystals under the compound fields was thoroughly investigated and explained.

2 EXPERIMENTAL PART

The monotectic alloy ingots with the composition of Cu-37.4w/%Pb were prepared by a mixture of pure metals with a purity of 99.99 w/%. The ingot preparations were carried out in a simulated microgravity equipment that designed by us. Figure 1 is a schematic diagram of the device.¹⁶ The temperature of the device was monitored by an infrared thermometer. The heating curve of the fabrication process was exemplified in Figure 2. The ingot prepared by normal solidification was heated to 1200 °C firstly, and then kept at 1200 °C for 60 min, followed by cooling to the room temperature. The ingot prepared from the intense pulse and electromagnetic compound fields had the same heating process. After the temperature was held for 60 min, an electromagnetic field (0.5 T) was applied for 2 min. In the cooling process at a temperature interval of 1200 °C to 955 °C, an intense pulse (with a width of 20 μ s, frequency of 30 Hz and peak current of 1800A) was employed. After that, the Cu-Pb ingot was cooled to room temperature and cut into small pieces for the microstructure and properties investigation.

The microstructures and compositions of the samples were investigated by scanning electron microscopy (SEM, SS-550 ISSEM), energy-dispersive spectroscopy (EDS, PHOENIX EDAX-2000) and X-ray diffraction (XRD, 6000X). Friction and wear experiments were performed on a HT-1000 friction-abrasion testing machine for 30 min at room temperature. The friction radius is 3 mm, the load is 5 N, and the rotation speed is 390 min⁻¹, samples were ground on GCr15 without lubrication. The microhardness tests were performed under a load of 4.9 N and a dwell time of 10 s.



Figure 2: Curve of heating temperature

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Figure 3: SEM of Cu-37.4w/%Pb, Cu-Pb alloy prepared under normal solidification (a and b) and complex fields (c and d). Transverse (a and c) and vertical (c and d) are the directions along and vertical to the current, respectively



Figure 4: Energy-dispersive spectroscopy (EDS) images of Cu-37.4*w*/%Pb monotectic alloy, Cu-Pb alloy prepared under normal solidification condition (a) and complex fields (b)

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3 RESULTS

3.1 Microstructure of Cu-37.4w/%Pb monotectic alloy

SEM images of the Cu-Pb monotectic alloy prepared under normal solidification and complex fields are shown in **Figure 3**, the EDS are shown in **Figure 4** and the XRD patterns of the alloy are shown in **Figure 5**.

As shown in **Figures 4** and **5**, a gray matrix denoted as α -Cu could be observed, and the Pb-phase could be



Figure 5: XRD patterns of the Cu-37.4w/%Pb monotectic alloy prepared under normal solidification conditions and complex fields

distinguished at the crystal grain boundaries of the Cu. The Cu-Pb alloy prepared by normal solidification is mainly composed of α -Cu and Pb with large crystal grains, in which the messily distributed Pb agglomerates on the matrix (**Figure 3a** and **3b**). With the help of the intense pulse and electromagnetic compound fields, the segregation almost disappears, as shown in **Figure 3c** and **3d**. The black-grid structure along the crystal grain boundary becomes lath-shaped, and the refined crystal grains radiationally grow from the guiding center to the surrounding area (**Figure 3d**).

3.2 Microhardness

In the microhardness testing, 7 points were randomly selected for each sample. The results are presented in **Table 1** (MAX T-L is the maximum difference).

As shown in **Table 1**, the average microhardness of Cu-Pb under normal solidification is $HV_{4.9}35.47$. Its microhardness is considerably increased to $HV_{4.9}58.50$ due to the treatment by complex fields. The MAX T-L is large and the hardness is not uniform.

Table 1: Microhardness test table of Cu-37.4w/%Pb (HV); MAX T-Lis the maximum difference

Test point	Normal solidification	Complex fields
1	37.88	57.10
2	36.24	57.07
3	35.65	61.24
4	42.29	57.49
5	33.80	55.30
6	35.84	63.37
7	26.58	57.90
Average	35.47	58.50
MAX T-L	15.71	8.07



Figure 6: Friction curves of the Cu-37.4w/%Pb monotectic alloy

The friction and wear properties of the alloy were evaluated using the friction coefficient. The variation of the friction coefficient with the time is shown in **Figure 6**. The average friction of coefficient and the weight loss (30 min) of the alloy are shown in **Table 2**.

It is obvious that the friction coefficient of the Cu-Pb alloy using normal solidification is stable after 10 minutes of testing, and its variation amplitude is larger compared with the one assisted by the complex field. Its average friction coefficient is 0.315, and the wear loss is 6.623 mg. The friction coefficient of the alloy prepared with complex fields needs 7 minutes for stabilization. Its average friction coefficient is 0.213, and the weight loss is substantially decreased by 51.5 %.

Table 2: Friction coefficient and weight loss of Cu-Pb alloy

Test sample	Normal solidification	Complex fields
Average friction coefficient	0.315	0.213
Abrasion loss (mg)	6.623	3.210

4 DISCUSSION

The Cu-Pb alloy is a kind of monotectic alloy. When the temperature of the Cu-Pb alloy is higher than the liquidus temperature, the liquid phase is completely miscible. As the liquid is cooled below to the liquidus temperature, the liquid phase separates into two phases $(L_1 + L_2)$. When the Cu-Pb monotectic alloy is cooled below to the monotectic reaction temperature of 955 °C, a monotectic reaction occurs:

$$L_1 \rightarrow \alpha(\mathrm{Cu}) + L_2$$
 (1)

The liquid phase L_1 decomposes into a solid phase α (Cu) and a liquid phase L_2 . The phase L_2 precipitates the surrounding α (Cu). As the temperature continues to decline, the redistribution of the solute occurs in both α (Cu) and L_2 phases. As a result, the rich-solute phase L_2 solidifies into a Pb-rich phase. As it is cooled to the eutectic temperature, the eutectic reaction occurs in the remaining liquid phase L_2 , forming two solid phases of Cu and Pb.

$$L_2 \rightarrow Cu + Pb$$
 (2)

Under normal solidification conditions the density difference between L_1 and L_2 in the Cu-Pb alloy is large, which leads to the phase stratification due to the Stockes deposition effect. As a result, severe phase segregation occurs due to gravity (**Figure 3a** and **3b**).

Without the electromagnetic field, the thermal movement of ions is random and arbitrary. As an electromagnetic field is applied (**Figure 7**), the velocity *V* of an ion is decomposed into two velocity components, i.e., perpendicular V_{\perp} and parallel V_{\parallel} . Along the vertical motion direction, the Lorentz force is produced because of the

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Figure 7: Magnetic effect under an energized coil

interaction between an ion with a charge of q and the electromagnetic field f:

$$\vec{\mathbf{f}} = q\vec{v}_1 \times \vec{\mathbf{B}} \tag{3}$$

The velocity V_{\perp} , without changes of the magnitude and the direction, is perpendicular to the Lorentz force. As a result, the ions rotate around the electromagnetic line with the load of the Lorentz force. Meanwhile, the force of V_{\parallel} results in the movement of the ions along the electromagnetic line. The influence of the electromagnetic field on a charged particle can be described by a motion Equation (4):

$$m\ddot{v} = q(\dot{v} \times B) \tag{4}$$

Where *m* is the mass of the charged particle, the Larmor circle transformation of Equation^{17,18} (4) is as follows in ¹⁹:

$$\begin{bmatrix} x - \left(x_0 - \frac{v_\perp}{\omega_c} \sin \alpha\right) \end{bmatrix}^2 + \begin{bmatrix} y - \left(y_0 - \frac{v_\perp}{\omega_c} \cos \alpha\right) \end{bmatrix}^2 =$$

$$= \frac{v_\perp^2}{\omega_c^2} = \frac{m^2 v_\perp^2}{q^2 B^2}$$
(5)

The particles move in the plane $z = v \parallel t + z_0$. Larmor's center is:

$$\left(x_{0} - \frac{v_{\perp}}{\omega_{c}}\sin\alpha, y_{0} - \frac{v_{\perp}}{\omega_{c}}\cos\alpha\right)$$

and the guiding center is:

$$r_{\rm c} = \left(x_0 - \frac{v_{\perp}}{\omega_{\rm c}} \sin \alpha, y_0 - \frac{v_{\perp}}{\omega_{\rm c}} \cos \alpha, v \parallel t + z_0 \right)$$

and the Larmor radius is:

$$r_c = \frac{v_\perp}{\omega_c} = \frac{mv_\perp}{|q|B}$$

The values of x_0 , y_0 , z_0 , V_{\perp} , V_{\parallel} and α are determined by the initial conditions. The Larmor radius, denoted as

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the cyclotron radius, is proportional to the ion mass and inversely proportional to the charge of q. Since the mass and charges of Cu⁺ and Pb⁺⁴ are considerably different, the size of their Larmor radii is obviously different. The relative movement of the solute particles under the action of the electromagnetic field enhances the diffusion of the solute Pb in Cu, which results in an increase of the solubility of Pb in Cu, and therefore, the suppression of the segregation (**Figure 3c** and **3d**). In the vertical direction of the pulse current, particles rotate in the plane



Figure 8: Cooling curves of Cu-37.4w/%Pb monotectic alloys

 $z=v_{\parallel}t+z_0$, where the crystal grains radiationally grow from the guiding center to the surrounding area (**Figure 3d**). In addition, the melt is repeatedly compressed via an electromagnetic force according to the magnetostrictive effect, which results in the rapid disappearance of the overheating in the melt.¹⁹ Thus the degree of undercooling is substantially increased. Meanwhile, the growth of the crystal grains is suppressed, and the dendrites are broken. As a result, the crystal grains are refined and they become lath-shaped.

According to the solidification theory, nucleation occurs in a melt under a certain condition of undercooling to overcome the interface energy between the liquid and the solid phases. The intense pulse current can effectively increase the undercooling degree of the melt during solidification, which enhances the nucleation. As a pulse current is applied into a melt, the influence of the current on the undercooling degree can be expressed as,

$$\Delta T = k J_0^2 \varepsilon \pi r^2 \Delta T_m C^{-1} \tag{6}$$

where ΔT is the melt undercooling, and J_0 is the current density before nucleation, and ΔV is the nucleation volume, and T_m is the melting point, and r is the spheroidal radius, and k is a constant associated with the spherical coordinates, and c is the mass heat capacity, and ε is the correlation coefficient of electroconductivity.^{19–21}

Figure 8 shows the cooling curves for the Cu-37.4*w*/%Pb monotectic alloys. With a calculation, the ΔT of Cu-Pb monotectic alloy substantially increases from 44 °C to 159 °C due to the treatment of complex fields, which is attributed to the enhancement of the nucleation and the refinement of the crystal grains.

Under normal solidification, the phase segregation is strong, and lots of large blocks of Pb are observed (**Figure 3a** and **3b**). The hardness difference between the Pb and the Cu-Pb alloy is considerable, which results in the obviously inhomogeneous distribution of the microhardness (**Table 1**). The improvement of the microhardness, due to the treatment of complex fields, results in the solution strengthening. The solid solubility of Pb in Cu is considerably increased. Furthermore, intense pulse results in the refinement of the crystal grains, which is attributed to the homogeneous distribution of the microhardness (**Table 1**).

The dissolution of Pb in the Cu matrix promotes the lubrication effect of the Cu-Pb alloy. Furthermore, the suppression of the phase separation and the refinement of crystals are beneficial to the wear resistance. Under the normal solidification, phase segregation is strong and the grains are coarse, where the friction coefficient is high and the amplitude of the friction coefficient is large (**Figure 6**). However, under the complex fields, the improvement of the microhardness and the refinement of crystals result in a considerably decrease of the friction coefficient and a small amplitude of the friction coefficient (**Figure 6**). Thus, the wear resistance of the alloy is greatly improved (**Table 2**).

5 CONCLUSIONS

1) Under normal solidification, phase segregation in the Cu-Pb monotectic alloy is strong, and the crystal grains are coarse, which leads to a low microhardness and poor wear resistance.

2) Under the intense pulse and electromagnetic complex fields, the phase separation in the Cu-Pb monotectic alloy is considerably suppressed, and the crystals are refined. The crystal grains radiationally grow from the guiding center to the surrounding area due to the gyroscopic motion caused by the Lorentz force.

3) Under the same friction condition, the hardness is considerably increased by 64.93 %, and the weight loss of the Cu-Pb alloy prepared by the complex fields is substantially reduced by 51.5 %, which indicates that the wear resistance is greatly improved.

Acknowledgements

This project was financially supported by the National Natural Science Foundation of China (Grant No. 51561001) and the State Key Laboratory of Special Rare Metal Materials.

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